SYNTHESES OF NEW 3-OXA-METHANO-PGI₁ DERIVATIVES AND THEIR BIOLOGICAL PROPERTIES

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SUMMARY: Syntheses of new PGI₁ derivatives and inhibitory activities of blood platelet aggregation are described. It is shown that high potent 3-oxa-methano-PGI₁ compound SM-10906 can be obtained, when the natural α-chain is modified by introducing a 3-oxa-moiety.

INTRODUCTION AND CONCEPT: Prostacyclin (PGI₂)¹ is a metabolite of arachidonic acid, which shows potent platelet aggregatory inhibitory and vasodilatory action. However, due to the labile enol ether moiety and intramolecular acid catalysis by the carboxylic moiety, prostacyclin is inactivated by being rapidly hydrolyzed in the blood.² This limits its possible clinical application and has led to the synthesis of physico-chemically stable and biochemically potent analogues of PGI₂. In these approaches, many stable PGI₂ derivatives³ having high biological activity were reported. The enol ether oxygen in PGI₂ was replaced by a methylene group, leading to potent methano-PGI₂ derivatives, such as Iloprost, OP-41483, Ciprostene, and Cicaprost, some of which are under clinical trials. Aza-PGI₂ derivatives, such as OP-2507, which incorporate an imino group, 5-cyano PGI₂ derivative, such as Nileprost, and benzo-PGI₂ derivatives, such as Beraprost and Taprostene, are also potent PGI₂ analogues. On the other hand, when the enol ether double bond in PGI₂ was replaced by a single bond, the PGI₁ derivatives were remarkably less potent than PGE₁ or PGI₂ in inhibiting the aggregation of blood platelets though physico-chemically stable. For example, exo-PGI₁ and endo-PGI₁ were respectively 20 and 400 times less potent than PGE₁, ³ and methano-PGI₁ was 80 times less potent than PGE₁.⁴

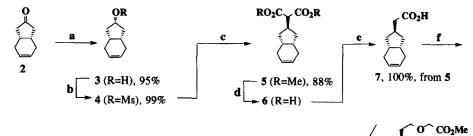
In our preliminary approaches, we tried to modify the methano-PGI₁ to increase the activity as well as physicochemical and metabolic stability by introducing a heteroatom into the natural α -chain of PGs, and found that the 3-oxa-methano-PGI₁ derivatives were substantially more potent than 4-oxa, 5-oxa-methano-PGI₁ derivatives⁶ in inhibitory activities of blood platelet aggregation. Further structural modification to increase the activity concentrated on the ω -chain of PGs and resulted in the synthesis of 3-oxa-17S,20-dimethyl-methano-PGI₁(1, SM-10906),⁷ which was the most potent compound of the 3-oxa-methano-PGI₁ derivatives. In this report, we describe synthetic procedures and biological activities of the 3-oxa-methano-PGI₁ derivative 1 and its isomers.

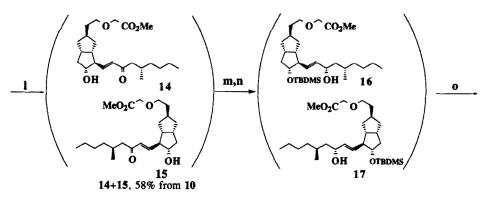
Stucture feature of SM-10906 in comparison with PGI₂

SYNTHESIS: Starting from the easily available ketone 2, we reduced the carbonyl group of 2 with Red Al under mild conditions (-65°C), resulting in α -alcohol 3 containing a small amount of β -alcohol (ca. 4%), which could be removed later. Mesylation of 3, followed by alkylation of mesylate with malonate, saponification of the diester 5 with KOH, and decarboxylation under reflux in xylene gave 3-β-carboxylic acid 7 (mp 52-53°C). A small amount of 3-α-carboxylic acid in the resultant acid 7 could be completely removed by repeated recrystallization from hexane. Reduction of 7 with Red Al, followed by alkylation with sodium chloroacetate and esterification gave the key intermediate 9. Epoxidation of the olefin in 9 in a two phase reaction system with hydrogen peroxide, alkali tungstate and phosphoric acid in the presence of quarternary ammonium salt, followed by ring-cleavage of the epoxide ring in 10 with periodic acid gave the labile dialdehyde 11. Aldol condensation of dialdehyde 11 in the presence of piperidine and acetic acid for 10 min and immediate Wittig-Horner reaction with dimethyl 4S-methyl-2-oxooctylphosphonate, followed by purification by chromatography gave a nearly 1:1 mixture of 14 and 15, which could not be separated by HPLC. After protection of the C-11 hydroxyl group in the mixture of 14 and 15 with TBDMSCl, reduction with NaBH4 in the presence of CeCl3⁸ at -45°C gave a mixture of 15-α-alcohols 16 and 17, and a small amount of their 15- β -isomers (84%yield, $\alpha/\beta=9/1$). The mixture of 15- α -alcohol 16 and 17 could be easily separated from their 15-β-isomers by chromatography. Though the desired alcohol 16 could not be separated by HPLC from 17. Therefore, we applied the Sharpless epoxidation for the purpose of effective isolation of 16 from 17. Sharpless epoxidation 10 of the mixture of 16 and 17 in the presence of (-)-DIPT, followed by deprotection gave the mixture of desired 20 and epoxide 19. Desired 20 could be easily separated from 19 by chromatography. The absolute configuration of 20 was confirmed by ¹H NMR and ¹³C NMR spectroscopy employing, Mosher's method 11 and line width method, and further supported by exciton coupled circular dichromic spectroscopy, and the configuration of the starting material 2.¹² Saponification of 20 with NaOH furnished SM-10906 (1).

The epimers of 1 were produced by the following procedures: Oxidation of 20 with MnO2, followed by reduction with NaBH4 gave a mixture of alcohol 20 and 21 which were readily separated by chromatography. Saponification of the ester 21 with NaOH furnished 22, the 15-epimer of 1. Mitsunobu reaction 13 of the alcohol group at the position 11 in enone 14, followed by reduction with NaBH4 gave a mixture, which were readily separated by chromatography to give the ester 24. Saponification of the ester 24 with NaOH furnished 25, the 11-epimer of 1. Wittig-Horner reaction of the mixture of 12 and 13 with dimethyl 4R-methyl-2-oxocotylphosphonate, followed by reduction with L-Selectride gave the mixture of alcohols, which were separated by HPLC to give the ester 28. Saponification of the ester 28 with NaOH furnished 29, the 17-epimer of 1.

Synthesis of SM-10906





Synthesis of 22

20 r
$$O \cap CO_2Me$$
 $O \cap CO_2R$ $O \cap CO_2R$

Synthesis of 29

CONDITIONS: a: Red Al, toluene, -65°C, 2hr; b: MsCl, triethylamine, toluene, 10°C, 1hr; c: NaH, dimethylmalonate, toluene, MeOH, 65°C, 3hr; d: 20%KOH, IPA, 80°C, 3hr; e: xylene, reflux, 5hr; f: Red Al, toluene, 45°C, 2hr; g: n-BuLi, DMSO, CICH2CO2Na, 50°C, 5hr; h: MeOH, c-H2SO4, 20°C, 15hr; i: H2O2, Na2WO4, 85% H3PO4, CH2Cl2, H2O, cetyltrimethylammonium bromide, 72°C, 5hr; j: HIO4, CH2Cl2, H2O, 25°C, 4hr; k: piperidine, acetic acid, -5°C, 10min; l: NaH, dimethyl 4S-methyl-2-oxooctylphosphonate, THF, rt, 2hr, SiO2; m: TBDMSCl, imidazole, DMF, rt, 5hr; n: NaBH4, CeCl3, 8 EtOH, -45°C, 8hr; o: (-) DIPT, Ti(Oi-Pr)4, TBHP, -20°C, 15hr; p: AcOH/MeOH/H2O=1/3/1, rt, 12hr, SiO2; q: 1N-NaOH, MeOH, 0°C, 30min; r: MnO2, chloroform, rt, 2hr, SiO2; s: NaBH4, MeOH, 0°C, 30min, SiO2; t: Ph3P, HCO2H, DEAD, THF, rt, 10hr; u: K2CO3, MeOH, rt, 2h, SiO2; v: NaH, dimethyl 4R-methyl-2-oxooctylphosphonate, THF, rt, 2hr, SiO2; w: L-Selectride, THF, -60°C, 30min, SiO2.

RESULTS: The biological data of PGI1 derivatives synthesized above are given in $Table\ 1,^{14,15}$ in which their inhibitory activities of blood platelet aggregation 16 in rabbit platelet-rich plasma stimulated by ADP were compared with PGE1 as standard. As depicted in $Table\ 1$, the activity of 1, SM-10906 was nearly equipotent with PGE1 and was remarkably enhanced in comparison with known PGI1 derivatives owing to introduction of 3-oxa moiety into natural α -chain. On the other hand, both PGI2 analogue, Iloprost and its 3-oxa-PGI2 derivative, Z96480 were high potent. The was 70 times more potent than methano-PGI1. (compound 1, A and B). As was expected in general, the stereoisomers of 1 were less potent compared with 1. For example, the 11-epimer and 15-epimer were over 300 times less potent than 1 and the 17-epimer was 3 times less potent (compound 1, 25, 22 and 29). In Table 2 and Table 3, 19 the platelet aggregation inhibitory activities of 1 induced by some aggregating agents in platelet-rich plasma of different species are provided. The agent 1 inhibited platelet aggregation induced by collagen and arachidonic acid as well as ADP and inhibited platelet aggregation in platelet-rich plasma of human, guinea-pig, dog and rat as well as rabbit. SM-10902 (20), which is a pro-drug of SM-10906 (1), is now under clinical study.

Table 1 Biological activities of SM-10906 and its isomers inhibition of rabbit blood platelet aggregation induced by ADP

Table 2 Biological activities of SM-10906 in various species inhibition of ADP-induced platelet aggregation (IC₅₀, ng/ml)

Species	SM-10906	PGE ₁
Human	12.5 ± 2.7	13.6 ± 2.7
Guinnea-pig	3.1 ± 0.5	2.5 ± 0.3
Dog	4.3 ± 0.7	3.8 ± 0.8
Rat	9.8 ± 0.7	34.4 ± 5.1
Rabbit	24.1 ± 4.6	20.7 ± 5.7

Table 3 Biological activity of SM-10906 with various aggregating reagents inhibition of rabbit platelet aggregation (IC₅₀, ng/ml)

aggregating reagent	SM-10906	PGE ₁
ADP	24.1 ± 4.6	20.7 ± 5.7
collagen	18.3 ± 4.4	13.2 ± 2.6
arachidonic acid	47.6 ± 4.7	35.9 ± 3.8

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- The C-15 hydroxyl unprotected enone in 14 and 15 was reduced by reducing reagent such as Yamamoto reagent, Noyori reagent, NaBH4, and L-Selectride to give the alcohol (respectively 18.0% yield, $\alpha/\beta = 1.1/1$; 19%yield, $\alpha/\beta = 2.2/1$; 26%yield, $\alpha/\beta = 1/2$; and 42%yield, $\alpha/\beta = 1.9/1$). The low α/β selectivity and relative lowyield of these cases may be due to the presence of 3-oxa moiety in 14 and 15. Because the C-15 unprotected enone without 3-oxa moiety was reduced by the same reducing reagents to give the alcohol in relatively high α/β selectivity and yield.
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